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212 Chemistry: Matter and Change • Chapter 11 Solutions Manual CHAPTER 11 SOLUTIONS MANUAL c. How many moles of H₂S are produced? 1.50 mol S 8 4 mol H₂S _ 1 mol S 8 6.00 mol H₂S 12. Challenge Sulfuric acid (H₂SO₄) is formed when sulfur dioxide (SO₂) reacts with oxygen and water. a. Write the balanced chemical equation for the reaction. 2SO₂(g) + O₂(g) → 2H₂SO₄(l)

StoichiometryStoichiometry

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The MoleThe Mole

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